```
2 S FERRIC SULFATE/CN OR FERRIC SUBSULFATE/CN
L1
              1 S ALUMINUM CHLORIDE/CN
L2
              2 S ALUMINUM AMMONIUM SULFATE/CN
L3
     FILE 'CAPLUS, WPIDS, MEDLINE, BIOSIS, USPATFULL' ENTERED AT 22:23:33 ON
     16 MAY 2004
     FILE 'REGISTRY' ENTERED AT 22:23:44 ON 16 MAY 2004
              SET SMARTSELECT ON
            SEL L1 1- CHEM: 17 TERMS
L4
                SET SMARTSELECT OFF
     FILE 'CAPLUS, WPIDS, MEDLINE, BIOSIS, USPATFULL' ENTERED AT 22:23:46 ON
     16 MAY 2004
         10534 S L4/BI
L5
     FILE 'REGISTRY' ENTERED AT 22:24:51 ON 16 MAY 2004
               SET SMARTSELECT ON
L6
            SEL L2 1- CHEM: 15 TERMS
                SET SMARTSELECT OFF
     FILE 'CAPLUS, WPIDS, MEDLINE, BIOSIS, USPATFULL' ENTERED AT 22:24:52 ON
     16 MAY 2004
L7
          70163 S L6/BI
     FILE 'REGISTRY' ENTERED AT 22:25:13 ON 16 MAY 2004
                SET SMARTSELECT ON
            SEL L3 1- CHEM: 19 TERMS
                SET SMARTSELECT OFF
     FILE 'CAPLUS, WPIDS, MEDLINE, BIOSIS, USPATFULL' ENTERED AT 22:25:14 ON
     16 MAY 2004
            15 S L10 AND L/ AND L9
15 S L10 AND (ROC OR ORC OR CELLULOS?)
12 DUP REM L11 (3 DUPLICATES REMOVED)

(1.e. not Just Regenerated cell.)
          11341 S L8/BI
L9
L10
L11
L12
=> d que
              2 SEA FILE=REGISTRY FERRIC SULFATE/CN OR FERRIC SUBSULFATE/CN
              1 SEA FILE=REGISTRY ALUMINUM CHLORIDE/CN
L2
              2 SEA FILE=REGISTRY ALUMINUM AMMONIUM SULFATE/CN
L3
                SEL L1 1- CHEM: 17 TERMS
L4
          10534 SEA L4/BI
L5
                SEL L2 1- CHEM: 15 TERMS
L6
          70163 SEA L6/BI
L7
                SEL L3 1- CHEM: 19 TERMS
L8
          11341 SEA L8/BI
L9
L10
           40 SEA L5 AND L7 AND L9
             15 SEA L10 AND (ROC OR ORC OR CELLULOS?)
L11
            12 DUP REM L11 (3 DUPLICATES REMOVED)
L12
```

FILE 'REGISTRY' ENTERED AT 22:21:54 ON 16 MAY 2004

```
=> d 1-12 bib ab
    ANSWER 1 OF 12 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 1
     2003:922612 CAPLUS
ΑN
DN
     139:386489
     Bleeding control and healing aid compositions and methods of use
TI
IN
     Prevendar, Terence
PA
SO
     U.S., 5 pp.
     CODEN: USXXAM
DT
     Patent
LA
    English
FAN.CNT 1
                                          APPLICATION NO.
    PATENT NO.
                     KIND DATE
                                          _____
                                                           -----
                                          US 2002-72772
                                                            20020208
                     В1
                           20031125
     US 6652840
PΙ
PRAI US 2001-270379P P
                           20010221
    A compn. is disclosed which has been shown to stop or control bleeding and
     seal open small blood vessels while accelerating the healing process of
     abraded oral "qum" and other "skin" (epithelial) tissues. The compn. is
     preferably in the form of a paste which promotes ease of application and
     use of the compn. A variety of instruments can be used in application and
     cleanup of the compn. showing versatile unparalleled friendly usage. The
     compn. preferably comprises aluminum chloride,
     ferric sulfate (subsulfate), regenerated oxidized
     cellulose, aluminum ammonium sulfate
     , absorbable gelatin and a solvent.
                                         The compn. has many dental and
     medical procedure applications. Following tooth or root extn. the compn.
     of the invention was placed over the optionally packed socket with gauze
     pressure to form a "bandage" with a suture. Healing time was reduced by
     four days. Patient experienced little discomfort and insignificant
     bleeding.
              THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT 17
              ALL CITATIONS AVAILABLE IN THE RE FORMAT
    ANSWER 2 OF 12 USPATFULL on STN
L12
       2002:275099 USPATFULL
ΑN
TΙ
       Package of sheet-type patches
       Ishida, Koichi, Tokyo, JAPAN
TN
       Kao Corporation, Tokyo, JAPAN (non-U.S. corporation)
PA
PΙ
       US 6467621
                         В1
                               20021022
       WO 9923012 19990514
                               19990826 (9)
ΑI
       US 1999-331579
                               19981030
       WO 1998-JP4927
                               19990826 PCT 371 date
                         19971031
       JP 1997-314600
PRAI
       Utility .
DT
       GRANTED
      Primary Examiner: Luong, Shian
EXNAM
       Oblon, Spivak, McClelland, Maier & Neustadt, P.C.
LREP
       Number of Claims: 5
CLMN
ECL
       Exemplary Claim: 1
       6 Drawing Figure(s); 5 Drawing Page(s)
DRWN
LN.CNT 535
       The present invention relates to a package of sheet-type patches (1)
AΒ
       which are applied to the face, and to a method for producing the same.
       In the invention, sheet-type patches (1) each comprising a pack agent
       layer (13) formed on a substrate (12) and covered with a liner layer
       (14) are packaged in a package pouch (19) in which said sheet-type
       patches (1) are fixed onto the inner surface of the package pouch (19)
```

via their liner layer (14). The sheet-type patches (1) can be fixed onto the inner surface of the package pouch (19) by means of an adhesive or

electrostatic force. According to the invention, the process of

producing sheet-type patches (1) and packaging them is much simplified, and the production efficiency in the process is much improved. In the package of sheet-type patches (1) of the invention, the patches do neither move nor adhere to each other.

```
L12 ANSWER 3 OF 12 USPATFULL on STN
       1999:160144 USPATFULL
ΑN
       Viscous carrier compositions, including gels, formed with an organic
TI
       liquid carrier, a layered material: polymer complex, and a di-, and/or
       tri-valent cation
       Tsipursky, Semeon, Lincolnwood, IL, United States
IN
       Dolinko, Vladimir, Libertyville, IL, United States
       Psihogios, Vasiliki, Elk Grove Village, IL, United States
       Beall, Gary W., McHenry, IL, United States
       Amcol International Corporation, Arlington Heights, IL, United States
PΑ
       (U.S. corporation)
       US 5998528
                               19991207
PΙ
ΑI
       US 1998-17421
                               19980202 (9)
       Continuation-in-part of Ser. No. US 1995-525416, filed on 8 Sep 1995,
RLI
       now patented, Pat. No. US 5721306 And Ser. No. US 1996-637092, filed on
       2 May 1996, now patented, Pat. No. US 5760121 which is a
       continuation-in-part of Ser. No. US 525416 Ser. No. Ser. No. US
       1995-488264, filed on 7 Jun 1995, now patented, Pat. No. US 5552469 And
       Ser. No. US 1995-488263, filed on 7 Jun 1995, now patented, Pat. No. US
       5698624 , said Ser. No. US 525416 which is a continuation-in-part of
       Ser. No. US 488264 which is a continuation-in-part of Ser. No. US 488263
       Ser. No. Ser. No. US 1995-480080, filed on 7 Jun 1995, now patented,
       Pat. No. US 5578672 And Ser. No. US 488263
DT
       Utility
FS
       Granted
       Primary Examiner: Seidleck, James J.; Assistant Examiner: Rajguru, U. K.
EXNAM
       Marshall, O'Toole, Gerstein, Murray & Borun
LREP
       Number of Claims: 90
CLMN
       Exemplary Claim: 1
ECL
DRWN
       19 Drawing Figure(s); 18 Drawing Page(s)
LN.CNT 3224
CAS INDEXING IS AVAILABLE FOR THIS PATENT.
       Intercalates formed by contacting the layer material, e.g., a
AB
       phyllosilicate, with an intercalant to sorb or intercalate the between
       adjacent platelets of the layered material. Sufficient intercalant
       polymer is sorbed between adjacent platelets to expand the adjacent
       platelets at least about 5 .ANG. (as measured after water removal to 5%
       by weight water), up to about 100 .ANG. and preferably in the range of
       about 10-45 .ANG., so that the intercalate easily can be exfoliated into
       individual platelets. A monovalent, divalent and/or trivalent cation is
       added to the intercalating composition, or after intercalation for
       surprising increases in viscosity. The intercalated complex is combined
       with an organic liquid into an unexpectedly viscous carrier material,
       for delivery of the carrier material, or for delivery of an active
       compound, e.g., a pharmaceutical, or cosmetic, or lubricant, e.g., food
       grade lubricants dissolved or dispersed in the carrier material.
       Alternatively, the intercalated complex can be exfoliated prior to
       combination with the organic liquid.
L12 ANSWER 4 OF 12 USPATFULL on STN
       1998:98615 USPATFULL
ΑN
ΤI
       Ruminant feed additive composition containing novel phosphoric
       acid-amino acid composite salt and water-soluble high-molecular
       substance
       Ikeda, Toru, Kawasaki, Japan
ΙN
```

Yukawa, Toshihide, Kawasaki, Japan

PA PI

AΙ

US 5795585

US 1996-777052

Ajinomoto Co., Inc., Tokyo, Japan (non-U.S. corporation)

19980818

19961230 (8)

JP 1995-343163 19951228 PRAI DT Utility FS Granted Primary Examiner: Levy, Neil S. EXNAM Oblon, Spivak, McClelland, Maier & Neustadt, P.C. LREP Number of Claims: 13 CLMN Exemplary Claim: 1 ECL DRWN No Drawings LN.CNT 954 CAS INDEXING IS AVAILABLE FOR THIS PATENT. A ruminant feed additive composition comprises a phosphoric acid-amino acid composite salt and a water-insoluble salt of a polyvalent-metalsensitive water-soluble high-molecular weight substance. The composite salt contains a basic amino acid, an alkaline-earth metal and phosphoric acid, and is insoluble in neutral or alkaline aqueous solution, but is soluble in acidic aqueous solution. ANSWER 5 OF 12 USPATFULL on STN L12ΑN 1998:65434 USPATFULL TIPhosphoric acid-amino acid-polyvalent metal composite salt and ruminant feed additive composition ΙN Hijiya, Toyoto, Kawasaki, Japan Ikeda, Toru, Kawasaki, Japan Mori, Kenichi, Kawasaki, Japan Yukawa, Toshihide, Kawasaki, Japan Takemoto, Tadashi, Kawasaki, Japan Kamada, Hajime, Kawasaki, Japan Ajinomoto Co., Inc., Tokyo, Japan (non-U.S. corporation) PΑ 19980609 PT US 5763657 WO 9724314 19970710 ΑT US 1997-894703 19970828 (8) WO 1996-JP3420 19961121 19970828 PCT 371 date 19970828 PCT 102(e) date PRAI JP 1995-343165 19951228 JP 1996-235309 19960905 DT Utility FS Granted Primary Examiner: Geist, Gary; Assistant Examiner: Keys, Rosalynd EXNAM LREP Oblon, Spivak, McClelland, Maier & Neustadt, P.C. CLMN Number of Claims: 21 ECL Exemplary Claim: 1 No Drawings DRWN LN.CNT 1535 CAS INDEXING IS AVAILABLE FOR THIS PATENT. The present specification discloses a ruminant feed additive composition which contains as an active ingredient a phosphoric acid-amino acid-polyvalent metal composite salt (final composite salt) which is insoluble in neutral or alkaline water and is soluble in acidic water and which can be obtained by treating a composite salt composed of a basic amino acid, magnesium and phosphoric acid with a salt of a divalent or trivalent (polyvalent) metal other than magnesium, or by treating the above-mentioned composite salt with the polyvalent metal salt and a condensed phosphoric acid component (alone) or the condensed phosphoric acid component and a phosphoric acid component (in combination), this composition taking the form of a powder or granules. The above-mentioned final composite salt can exhibit the excellent stability to neutral or slightly acidic water, namely, the low solubility therein in comparison with the intermediate composite salt, and it can have both the excellent insolubility of the basic amino acid in a rumen of a ruminant and the excellent elution thereof in an

abomasum and lower digestive organs.

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1998:44921 USPATFULL
AN
       Ruminant feed additive composition containing novel phosphoric
TI
       acid-amino acid-polyvalent metal composite salt and gastric antacid
       Ikeda, Toru, Kawasaki, Japan
ΤN
       Yukawa, Toshihide, Kawasaki, Japan
       Kobayashi, Hisamine, Kawasaki, Japan
       Sato, Hiroyuki, Kawasaki, Japan
       Kitamura, Nobuyoshi, Kawasaki, Japan
       Ajinomoto Co., Inc., Tokyo, Japan (non-U.S. corporation)
PA
                               19980428
       US 5744178
PΙ
       US 1996-774362
                               19961227 (8)
AΙ
PRAI
       JP 1995-341250
                           19951227
DT
       Utility
       Granted
FS
      Primary Examiner: Paden, Carolyn
EXNAM
       Oblon, Spivak, McClelland, Maier & Neustadt, P.C.
LREP
       Number of Claims: 21
CLMN
ECL
       Exemplary Claim: 1
DRWN No Drawings
LN.CNT 1539
CAS INDEXING IS AVAILABLE FOR THIS PATENT.
       A ruminant feed additive composition in the form of a powder or granules
       comprises as active ingredients a gastric antacid, a phosphoric
       acid-amino acid-polyvalent metal composite salt and optionally a
       water-insoluble salt of a polyvalent metal-sensitive water-soluble
       high-molecular weight substance. The phosphoric acid-amino
       acid-polyvalent metal composite salt is composed of a phosphoric acid
       such as orthophosphoric acid, a polyphosphoric acid or a metaphosphoric
       acid, a basic amino acid such as lysine or methionine, an alkaline earth
       metal such as calcium or magnesium and, optionally, a polyvalent metal
       such as aluminum.
L12 ANSWER 7 OF 12 USPATFULL on STN
       94:82343 USPATFULL
ΑN
       Method of production of polyvinyl chloride resin for paste processing
TI
       Nishina, Masaaki, Yokohama, Japan
ΙN
       Ozaki, Osamu, Himi, Japan
       Nippon Zeon Co., Ltd., Tokyo, Japan (non-U.S. corporation)
PΑ
                               19940920
PΙ
       US 5349049
       US 1993-40192
                               19930331 (8)
ΑI
       JP 1992-105770
                           19920331
PRAI
       Utility
DT
FS
       Granted
       Primary Examiner: Schofer, Joseph L.; Assistant Examiner: Weber, Tom
EXNAM
       Millen, White, Zelano & Branigan
LREP
       Number of Claims: 16
CLMN
       Exemplary Claim: 1
ECL
       No Drawings
DRWN
LN.CNT 630
CAS INDEXING IS AVAILABLE FOR THIS PATENT.
       A method of production of polyvinyl chloride resin for paste processing
AB
       comprises recovering in 98% or more the polyvinyl chloride resin from an
       aqueous dispersion of the polyvinyl chloride resin for paste processing
       as aggregates by adding an organic fluid which is at most barely soluble
       in water and does not dissolve or swell the polyvinyl chloride resin to
       the aqueous dispersion in the presence of an aggregating agent, followed
       by separating the aggregated polyvinyl chloride resin particles from the
       aqueous phase of the aqueous dispersion. By the addition of the
       aggregating agent, dispersion of the resin particles into a medium is
       improved, fluidity of a sol thereof and the physical properties of
       molded articles formed therefrom are improved and blocking during the
       drying process is prevented.
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1990:25649 CAPLUS
AN
DN
     112:25649
     Poultice containing ionomers in the carrier
TΙ
     Kusano, Takashi; Ishisone, Hiroyuki; Ozeki, Iwao
ΙN
     Horiuchi Itaro Co., Ltd., Japan
PΑ
     Jpn. Kokai Tokkyo Koho, 6 pp.
SO
     CODEN: JKXXAF
     Patent
DT
     Japanese
LA
FAN.CNT 1
                                           APPLICATION NO.
                                                            DATE
     PATENT NO.
                      KIND DATE
                      ____
                            _____
                                           JP 1987-185508
                                                            19870727
                            19890131
PI
     JP 01029306
                      Α2
                            19870727
PRAI JP 1987-185508
    A water-contg. poultice is prepd. by applying a paste contg. water-sol.
     polymer and a pharmaceutical to the surface of cloth contg. a water-sol.
     metal salt which produces a metallic ionomer upon reacting with the
     polymer. The leaking and oozing of the paste through the poultice compn.
     are prevented in this formulation. Thus, a paste was prepd. consisting of
     kaolin 10, poly(Na acrylate) 4, CM cellulose 2, a carboxyvinyl
     polymer 0.3, concd. glycerin 30, gelatins 3, Na aluminate metasilicate
     0.1, Me salicylate 1, dl-camphor 1, l-menthol 0.5, thymol 0.1, and H2O 48
     parts by wt. This paste (209) was applied to the surface (140 cm2) of
     cloth contq. AlK(SO4)2.
    ANSWER 9 OF 12 USPATFULL on STN
L12
       86:71459 USPATFULL
AN
       Hydrothermally treated product of compound having hydrotalcite-like
TΙ
       crystal structure composition thereof, and use thereof
       Miyata, Shigeo, Takamatsu, Japan
IN
       Anabuki, Hitoshi, Takamatsu, Japan
       Kyowa Chemical Industry Co., Ltd., Tokyo, Japan (non-U.S. corporation)
PΑ
       US 4629626
                               19861216
PI
                               19840627 (6)
       US 1984-625244
ΑI
       20001115
DCD
       JP 1983-114364
                           19830627
PRAI
DT
       Utility
       Granted
FS
       Primary Examiner: Waddell, Frederick E.
EXNAM
       Sherman and Shalloway
LREP
       Number of Claims: 12
CLMN
ECL
       Exemplary Claim: 1
       1 Drawing Figure(s); 1 Drawing Page(s)
CAS INDEXING IS AVAILABLE FOR THIS PATENT.
       A hydrothermally treated product useful for treating an iron deficiency
       of a compound having a hydrotalcite-like crystal structure and
       represented by the following formula (I)
       (Fe.sup.2+,Mg.sup.2+).sub.x (Al.sup.3+,Fe.sup.3+).sub.2 (OH).sub.2x+6-nz
       (A.sup.n-).sub.z.mH.sub.2 O
       wherein A.sup.n- represents an anion having a valence of n, Mg.sup.2+ is
       within the range of O.ltoreq.Mg.sup.2+ <x, Fe.sup.3+ is within the range
       of O.ltoreq.Fe.sup.3+ <2, x is a number represented by 1.ltoreq.x<20, z
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L12 ANSWER 10 OF 12 USPATFULL on STN

AN 83:53427 USPATFULL

TI Composition and method for treating iron deficiency syndrome

about 200.degree. C. in an aqueous medium.

is a number represented by 0<z<3, and m is a number represented by 0.1×20 , the product being formed by the thermal treatment of the compound of formula (I) at a temperature of about 100.degree. C. to

IN Anabuki, Hitoshi, Takamatsu, Japan

Miyata, Shigeo, Takamatsu, Japan Kyowa Chemical Industry C. Ltd., Tokyo, Japan (non-U.S. corporation) PΑ 19831115 US 4415555 PΤ 19820317 (6) US 1982-359126 ΑT 19810324 JP 1981-41723 PRAI DT Utility Granted FS Primary Examiner: Robinson, Douglas W. EXNAM Sherman & Shalloway LREP Number of Claims: 18 CLMN ECL Exemplary Claim: 1 DRWN 1 Drawing Figure(s); 1 Drawing Page(s) LN.CNT 634 CAS INDEXING IS AVAILABLE FOR THIS PATENT. A composition for treating an iron deficiency disease, composed of (1) an amount, effective for treating said disease, of a compound having a hydrotalcite-like crystal structure and represented by the following formula (Fe.sup.2+,Mg.sup.2+).sub.x (Al.sup.3+,Fe.sup.3+).sub.2 (OH).sub.2x+6-nz (A.sup.n-).sub.z.mH.sub.2 0 wherein A.sup.n- represents an anion having a valence of n, Mg.sup.2+ is within the range of O.ltoreq.Mg.sup.2+ <x, Fe.sup.3+ is within the range of O.ltoreq.Fe.sup.3+ 2, x is a number represented by 1.ltoreq..times.<20, z is a number represented by 0<z<3, and m is a number represented by O.ltoreq.m<8, and (2) a pharmaceutically acceptable diluent or carrier, and a method for treating an iron deficiency disease, which comprises orally administering an amount, effective for treating said disease, of the compound represented by the aforesaid formula. L12 ANSWER 11 OF 12 USPATFULL on STN 79:36568 USPATFULL ΑN Styptic composition TΙ Brown, Robert, 3249 Greenfield Dr., Marietta, GA, United States 30067 TN Setloff, Jerome, 6851 Roswell Rd., Apt. A 14, Atlanta, GA, United States 30328 19790828 US 4166108 PΙ US 1977-825853 19770818 (5) ΑI Continuation-in-part of Ser. No. US 1977-764405, filed on 31 Jan 1977, RLTnow abandoned DTUtility Granted. FS Primary Examiner: Moyer, Donald B. EXNAM Burns, Doane, Swecker & Mathis LREP Number of Claims: 12 CLMN Exemplary Claim: 1,11,12 ECL No Drawings DRWN LN.CNT 374 CAS INDEXING IS AVAILABLE FOR THIS PATENT. An improved styptic composition which effectively curtails bleeding while administering a soothing sensation on the injured area. The styptic composition is in the form of a stable lotion or cream having the following general formula: (1) from about 2% to about 12% by weight of a long chain fatty acid;

- (2) from about 0.4% to about 3.5% by weight of a wax filler;
- (3) from about 1% to about 10% by weight of polyethylene glycol stearate;
- (4) from about 1% to about 5% by weight of polyethylene glycol sorbitan

beeswax; and,

- (5) from about 0.5% to about 25% by weight of an acidic metallic salt,
- (6) the balance being water.

Optionally, from about 1% to about 4% by weight of glycerin may be added as a humectant.

Any conventional antiseptic may also be added to the composition.

The styptic composition may be packaged in any convenient form including jars, bottles, tubes, pump applicators, aerosol canisters and can also be impregnated into pre-packaged bandages.

L12 ANSWER 12 OF 12 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1976:410053 CAPLUS

DN 85:10053

TI Waste water purification

IN Hoeltgen, James B.; Humphrey, Harold E. B.

PA Kelmik, Inc., USA

SO U.S., 8 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI US 3928195	A	19751223	US 1974-512614	19741007
PRAI US 1972-317798		19721222		2

AB Human waste sludge, paper mill lagoon sludge, and biol. and chem. pharmaceutical sludges were purified in presence of sol. aluminates, silicates, and sol. cation sources. Thus, 100 ml of human waste sludge having PO42- 3300, NH4+ 390, and BOD 100,000-400,000 ppm was stirred at 200 rpm and pH >5.5 with Na aluminate [1302-42-7] 265, Na silicate [1344-09-8] 261, and CaCl2 200 mg and filtered to give an effluent having PO42- 0, NH4+ 100, and BOD 100 ppm.

=> d his

L1

T.4

(FILE 'HOME' ENTERED AT 22:21:42 ON 16 MAY 2004)

FILE 'REGISTRY' ENTERED AT 22:21:54 ON 16 MAY 2004

2 S FERRIC SULFATE/CN OR FERRIC SUBSULFATE/CN

L2 1 S ALUMINUM CHLORIDE/CN

L3 2 S ALUMINUM AMMONIUM SULFATE/CN

FILE 'CAPLUS, WPIDS, MEDLINE, BIOSIS, USPATFULL' ENTERED AT 22:23:33 ON 16 MAY 2004

FILE 'REGISTRY' ENTERED AT 22:23:44 ON 16 MAY 2004

SET SMARTSELECT ON

SEL L1 1- CHEM: 17 TERMS
SET SMARTSELECT OFF

FILE 'CAPLUS, WPIDS, MEDLINE, BIOSIS, USPATFULL' ENTERED AT 22:23:46 ON

16 MAY 2004 L5 10534 S L4/BI

FILE 'REGISTRY' ENTERED AT 22:24:51 ON 16 MAY 2004

SET SMARTSELECT ON

L6 SEL L2 1- CHEM: 15 TERMS SET SMARTSELECT OFF

```
FILE 'CAPLUS, WPIDS, MEDLINE, BIOSIS, USPATFULL' ENTERED AT 22:24:52 ON
     16 MAY 2004
ь7
          70163 S L6/BI
     FILE 'REGISTRY' ENTERED AT 22:25:13 ON 16 MAY 2004
                SET SMARTSELECT ON
L8
            SEL L3 1- CHEM:
                                19 TERMS
                SET SMARTSELECT OFF
     FILE 'CAPLUS, WPIDS, MEDLINE, BIOSIS, USPATFULL' ENTERED AT 22:25:14 ON
     16 MAY 2004
          11341 S L8/BI
L9
             40 S L5 AND L7 AND L9
L10
             15 S L10 AND (ROC OR ORC OR CELLULOS?)
L11
             12 DUP REM L11 (3 DUPLICATES REMOVED)
L12
=> d 2-5 11 bib ab kwic
   ANSWER 2 OF 12 USPATFULL on STN
       2002:275099 USPATFULL
ΑN
TΙ
       Package of sheet-type patches
       Ishida, Koichi, Tokyo, JAPAN
ΤN
       Kao Corporation, Tokyo, JAPAN (non-U.S. corporation)
PA
                               20021022
PΤ
       US 6467621
                          В1
       WO 9923012 19990514
       US 1999-331579
                               19990826 (9)
ΑT
       WO 1998-JP4927
                               19981030
                               19990826 PCT 371 date
       JP 1997-314600
                           19971031
PRAI
DΨ
       Utility
FS
       GRANTED
       Primary Examiner: Luong, Shian
EXNAM
       Oblon, Spivak, McClelland, Maier & Neustadt, P.C.
LREP
CLMN
       Number of Claims: 5
ECL
       Exemplary Claim: 1
DRWN
       6 Drawing Figure(s); 5 Drawing Page(s)
LN.CNT 535
       The present invention relates to a package of sheet-type patches (1)
AΒ
       which are applied to the face, and to a method for producing the same.
       In the invention, sheet-type patches (1) each comprising a pack agent
       layer (13) formed on a substrate (12) and covered with a liner layer
       (14) are packaged in a package pouch (19) in which said sheet-type
       patches (1) are fixed onto the inner surface of the package pouch (19)
       via their liner layer (14). The sheet-type patches (1) can be fixed onto
       the inner surface of the package pouch (19) by means of an adhesive or
       electrostatic force. According to the invention, the process of
       producing sheet-type patches (1) and packaging them is much simplified,
       and the production efficiency in the process is much improved. In the
       package of sheet-type patches (1) of the invention, the patches do
       neither move nor adhere to each other.
DETD
        . . metals and containing water, oily components and others;
       film-forming compounds such as polyvinyl alcohol, polyvinyl pyrrolidone,
       polyvinyl acetate emulsion, carboxymethyl cellulose, along
       with a vehicle of an inorganic powder of kaolin, talc, bentonite,
       titanium dioxide, zinc oxide and the like.
       . . . metal salts used as the crosslinking agents for the
DETD
       water-soluble polymer compounds, which include, for example, calcium
       chloride, magnesium chloride, aluminium chloride,
       potash alum, ammonium alum, iron alum,
       aluminum sulfate, ferric sulfate, magnesium sulfate
       and the like.
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1999:160144 USPATFULL ΑN Viscous carrier compositions, including gels, formed with an organic TΙ liquid carrier, a layered material: polymer complex, and a di-, and/or tri-valent cation Tsipursky, Semeon, Lincolnwood, IL, United States IN Dolinko, Vladimir, Libertyville, IL, United States Psihogios, Vasiliki, Elk Grove Village, IL, United States Beall, Gary W., McHenry, IL, United States Amcol International Corporation, Arlington Heights, IL, United States PΑ (U.S. corporation) 19991207 PΙ US 5998528 19980202 (9) ΑI US 1998-17421 Continuation-in-part of Ser. No. US 1995-525416, filed on 8 Sep 1995, RLI now patented, Pat. No. US 5721306 And Ser. No. US 1996-637092, filed on 2 May 1996, now patented, Pat. No. US 5760121 which is a continuation-in-part of Ser. No. US 525416 Ser. No. Ser. No. US 1995-488264, filed on 7 Jun 1995, now patented, Pat. No. US 5552469 And Ser. No. US 1995-488263, filed on 7 Jun 1995, now patented, Pat. No. US 5698624 , said Ser. No. US 525416 which is a continuation-in-part of Ser. No. US 488264 which is a continuation-in-part of Ser. No. US 488263 Ser. No. Ser. No. US 1995-480080, filed on 7 Jun 1995, now patented, Pat. No. US 5578672 And Ser. No. US 488263 Utility DTGranted FS Primary Examiner: Seidleck, James J.; Assistant Examiner: Rajguru, U. K. EXNAM Marshall, O'Toole, Gerstein, Murray & Borun LREP Number of Claims: 90 CLMN Exemplary Claim: 1 ECL 19 Drawing Figure(s); 18 Drawing Page(s) DRWN LN.CNT 3224 CAS INDEXING IS AVAILABLE FOR THIS PATENT. Intercalates formed by contacting the layer material, e.g., a phyllosilicate, with an intercalant to sorb or intercalate the between adjacent platelets of the layered material. Sufficient intercalant polymer is sorbed between adjacent platelets to expand the adjacent platelets at least about 5 .ANG. (as measured after water removal to 5% by weight water), up to about 100 .ANG. and preferably in the range of about 10-45 .ANG., so that the intercalate easily can be exfoliated into

platelets at least about 5 .ANG. (as measured after water removal to 5% by weight water), up to about 100 .ANG. and preferably in the range of about 10-45 .ANG., so that the intercalate easily can be exfoliated into individual platelets. A monovalent, divalent and/or trivalent cation is added to the intercalating composition, or after intercalation for surprising increases in viscosity. The intercalated complex is combined with an organic liquid into an unexpectedly viscous carrier material, for delivery of the carrier material, or for delivery of an active compound, e.g., a pharmaceutical, or cosmetic, or lubricant, e.g., food grade lubricants dissolved or dispersed in the carrier material. Alternatively, the intercalated complex can be exfoliated prior to combination with the organic liquid.

DETD

Aluminum

Aluminum acetate oxide

Aluminum ammonium sulfate

Aluminum antimonide
Aluminum arsenide
Aluminum boride
Aluminum bromide
Aluminum n-butoxide
Aluminum sec-butoxide
Aluminum sec-butoxide stearate
Aluminum t-butoxide

Aluminum carbide

Aluminum chloride
Aluminum di (sec-butoxide) acetoacetic

ester chelate
Aluminum di(isopropoxide)acetoacetic

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ester chelate
  Aluminum fluoride
  Aluminum hydroxide
  Aluminum iodide
  Aluminum isopropoxide
  Aluminum metaphosphate
  Aluminum molybdenum.
       . . . perchlorate
  Iron (III) perchlorate
  Iron (III) phosphate
  Iron phosphide
  Iron (III) pyrophosphate
  Iron (II) selenide
  Iron silicide
  Iron (II) sulfate
    Iron (III) sulfate
  Iron (II) sulfide
  Iron (IV) sulfide
  Iron telluride
  Iron (II) tetrafluoroborate
  Iron (III) meso-tetraphenylporphine, chlorine tree
  Iron (II) titanium oxide
     . hexacyanoferrate (III)
  Potassium trioxalatoferrate (III)
  (R) - (S) - PPFA
  Sodium hexafluoroiron (III)
  Sodium iron oxide
  Sodium pentacyanonitrosylferrate (II)
  Strontium dodecairon nonadecaoxide
  Tetraethylammonium(.mu.oxo)bis(trichloroferrate
  Tris(ethylenediamine)iron (III) sulfate
  Vinylferrocene ·
  Zinc iron oxide
  Diphenyllead dichloride
  Hexaphenyldilead
  Lead (II) acetate
  Lead (IV) acetate
  Lead (II) bromide
  Lead (II) carbonate
              other metal salts of olefin/maleic acid copolymers; sodium
DETD
       polymethacrylate; sodium polystyrene sulfonate; sodium
       styrene/acrylate/PEG-10 dimaleate copolymer; water-soluble esters and
       ethers of cellulose; sodium styrene/PEG-10
       maleate/nonoxynol-10 maleate/acrylate copolymer;
       starch/acrylate/acrylamide copolymers; styrene/acrylamide copolymer;
       styrene/acrylate/ammonium methacrylate copolymer; styrene/maleic
       anhydride copolymer; styrene/PVO copolymer; sucrose benzoate/sucrose
       acetate.
       . . . zinc-neomycin sulfate-hydrocortisone, chloramphenicol,
DETD
       methylbenzethonium chloride, and erythromycin and the like;
       antiparasitics, such as lindane; deodorants, such as chlorophyllin
       copper complex, aluminum chloride, aluminum
       chloride hexahydrate, and methylbenzethonium chloride;
       essentially all dermatologicals, like acne preparations, such as benzoyl
       peroxide, erythromycin-benzoyl peroxide, clindamycin phosphate,
       5,7-dichloro-8-hydroxyquinoline, and.
L12 ANSWER 4 OF 12 USPATFULL on STN
AN
       1998:98615 USPATFULL
       Ruminant feed additive composition containing novel phosphoric
TΙ
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acid-amino acid composite salt and water-soluble high-molecular

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substance
       Ikeda, Toru, Kawasaki, Japan
IN
       Yukawa, Toshihide, Kawasaki, Japan
       Ajinomoto Co., Inc., Tokyo, Japan (non-U.S. corporation)
PΑ
                               19980818
       US 5795585
PΙ
       US 1996-777052
                               19961230 (8)
AΙ
       JP 1995-343163
                           19951228
PRAI
DT
       Utility
FS
       Granted
       Primary Examiner: Levy, Neil S.
EXNAM
       Oblon, Spivak, McClelland, Maier & Neustadt, P.C.
LREP
       Number of Claims: 13
CLMN
       Exemplary Claim: 1
ECL
DRWN
       No Drawings
LN.CNT 954
CAS INDEXING IS AVAILABLE FOR THIS PATENT.
       A ruminant feed additive composition comprises a phosphoric acid-amino
AΒ
       acid composite salt and a water-insoluble salt of a polyvalent-metal-
       sensitive water-soluble high-molecular weight substance. The composite
       salt contains a basic amino acid, an alkaline-earth metal and phosphoric
       acid, and is insoluble in neutral or alkaline aqueous solution, but is
       soluble in acidic aqueous solution.
       . . high-molecular weight substances and proteins. Examples of the
SUMM
       polysaccharides having the carboxyl group include alginic acid, gellan
       gum, pectin, carboxymethyl cellulose and carboxymethyl starch.
       Examples of the high-molecular weight substances include polyacrylic
       acid and/or a copolymer of polyacrylic acid, and polymethacrylic. .
       . . . the polyvalent metal ion other than magnesium is used. Specific
SUMM
       examples thereof include aqueous solutions of aluminum salts such as
       aluminum chloride, polyaluminum chloride, aluminum
       sulfate, ammonium alum and potassium alum; aqueous
       solutions or aqueous dispersions of calcium salts such as calcium
       chloride, calcium sulfate, calcium hydroxide and calcium nitrate;
       aqueous solutions of iron salts such as ferrous chloride, ferric
       chloride, ferrous sulfate, ferric sulfate, potassium
       iron sulfate and ammonium iron sulfate; and aqueous solutions or aqueous
       dispersions of zinc salts such as zinc chloride,.
       . . . water-soluble binders and hydrophobic binders. Specific
SUMM
       examples of the water-soluble binders include water-soluble
       polysaccharides such as a starch, a carboxymethyl cellulose
       salt, an alginate, hydroxypropyl cellulose and a starch
       glycolic acid salt; water-soluble proteins such as casein sodium,
       gelatin and soybean protein; saccharides such as molasses,. .
       vegetable fats and oils; nonionic surfactants such as glycerin
       monostearate; and semi-synthetic resins and synthetic high-molecular
       substances such as acetyl cellulose, polyvinyl acetate, ester
       gum and a coumarone resin.
       . . . of alkaline-earth metals, talc, clay, bentonite and fine
SUMM
       silica; and organic substances such as paraffin wax, polyethylene
       powder, pulp powder, cellulose powder and xanthone.
         . . the solid material was separated from the mixed solution
DETD
       through filtration, 300 ml of water and 3.0 g of carboxymethyl
       cellulose sodium salt were added. These were mixed well, and the
       mixture was dried to obtain from 252 g to 241 g of each of desired
       compositions-1 to -3 comprising the composite salt and carboxymethyl
       cellulose calcium salt.
       . . . the intermediate starting composite salt-7 obtained in Example
DETD
       7 were mixed with 1,000 ml of water, and 30 g of {\tt ammonium}
       aluminum sulfate (burnt alum) were added. The mixture
       was stirred at room temperature for 2 hours. After the solid material
       was separated. .
                                            . . rumen in
DETD
adminis-
tering small
```

```
amount
Elution into
      100% 100%
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100% 100% 100% 100%

100%

abomasum

Composition No. 8 - 18-2 8 - 39-4 *1

carboxymethyl

carboxymethyl

carboxymethyl

calcium

cellulose Ca salt

cellulose Ca

cellulose Ca

alginate

salt salt Lysine content 46.0 18.1 17.0 18.4 Mg content 15.7 14.7 15.0 Polyvalent metal other Ca Ca Ca

than.

DETD Two-hundred grams of composition-1 comprising the composite salt and carboxymethyl cellulose calcium salt as obtained in Example 8 were mixed with 150 g of a hardened soybean oil. Then, the mixture.

CLM What is claimed is:

- calcium salt, an aluminum salt, a zinc salt and an iron salt of alginic acid, carrageenan, gellan gum, pectin, carboxymethyl cellulose, carboxymethyl starch, polyacrylic acid, a polyacrylic acid copolymer, polymethacrylic acid, a polymethacrylic acid copolymer, soybean protein or casein, and (ii).
- polymeric substance is at least one member selected from the group consisting of alginic acid, carrageenan, gellan gum, pectin, carboxymethyl cellulose, carboxymethyl starch, polyacrylic acid, a polyacrylic acid copolymer, polymethacrylic acid, a polymethacrylic acid copolymer, soybean protein or casein, and drying.
- polymeric substance is at least one member selected from the group consisting of alginic acid, carrageenan, gellan gum, pectin, carboxymethyl cellulose, carboxymethyl starch, polyacrylic acid, a polyacrylic acid copolymer, polymethacrylic acid, a polymethacrylic acid copolymer, soybean protein or casein, and contacting.
- ANSWER 5 OF 12 USPATFULL on STN I.12
- AN 1998:65434 USPATFULL
- TΙ Phosphoric acid-amino acid-polyvalent metal composite salt and ruminant feed additive composition
- IN Hijiya, Toyoto, Kawasaki, Japan Ikeda, Toru, Kawasaki, Japan Mori, Kenichi, Kawasaki, Japan Yukawa, Toshihide, Kawasaki, Japan Takemoto, Tadashi, Kawasaki, Japan Kamada, Hajime, Kawasaki, Japan
- PAAjinomoto Co., Inc., Tokyo, Japan (non-U.S. corporation)
- 19980609 PT US 5763657

WO 9724314 19970710

US 1997-894703 19970828 (8) ΑI WO 1996-JP3420 19961121

19970828 PCT 371 date 19970828 PCT 102(e) date

PRAI JP 1995-343165

19951228 JP 1996-235309 19960905

DTUtility FS Granted

Primary Examiner: Geist, Gary; Assistant Examiner: Keys, Rosalynd EXNAM

Oblon, Spivak, McClelland, Maier & Neustadt, P.C. LREP

CLMN Number of Claims: 21 Exemplary Claim: 1 ECL

No Drawings DRWN

LN.CNT 1535

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

The present specification discloses a ruminant feed additive composition which contains as an active ingredient a phosphoric acid-amino acid-polyvalent metal composite salt (final composite salt) which is insoluble in neutral or alkaline water and is soluble in acidic water and which can be obtained by treating a composite salt composed of a basic amino acid, magnesium and phosphoric acid with a salt of a divalent or trivalent (polyvalent) metal other than magnesium, or by treating the above-mentioned composite salt with the polyvalent metal salt and a condensed phosphoric acid component (alone) or the condensed phosphoric acid component and a phosphoric acid component (in combination), this composition taking the form of a powder or granules. The above-mentioned final composite salt can exhibit the excellent stability to neutral or slightly acidic water, namely, the low solubility therein in comparison with the intermediate composite salt, and it can have both the excellent insolubility of the basic amino acid in a rumen of a ruminant and the excellent elution thereof in an abomasum and lower digestive organs.

SUMM

. . solution, of the polyvalent metal ion other than magnesium. Specific examples thereof include aqueous solutions of aluminum salts such as aluminum chloride, polyaluminum chloride, ammonium sulfate, ammonium alum and potassium alum; aqueous solutions or aqueous dispersions of calcium salts such as calcium chloride, calcium sulfate, calcium hydroxide and calcium nitrate; aqueous solutions of iron salts such as ferrous chloride, ferric chloride, ferrous sulfate, ferric sulfate, potassium iron sulfate and ammonium iron sulfate; and aqueous solutions or aqueous dispersions of zinc salts such as zinc chloride,.

SUMM

Specific examples of the substance of the divalent or trivalent (polyvalent) metal other than magnesium include aluminum salts such as aluminum chloride, polyaluminum chloride, aluminum sulfate, ammonium alum and potassium alum; calcium salts or hydroxides such as calcium chloride, calcium sulfate, calcium hydroxide and calcium nitrate; iron salts such as ferrous chloride, ferric chloride, ferrous sulfate, ferric sulfate, potassium iron sulfate and ammonium iron sulfate; and zinc salts such as zinc chloride and ammonium zinc chloride; and zinc. . .

SUMM

. . . water-soluble binders and hydrophobic binders. Specific examples of the water-soluble binders include water-soluble polysaccharides such as a starch, a carboxymethyl cellulose salt, an alginate, hydroxypropyl cellulose and a starch glycolic acid salt; water-soluble proteins such as casein sodium, gelatin and soybean protein; saccharides such as molasses,. . vegetable fats and oils; nonionic surfactants such as qlycerin monostearate; and semi-synthetic resins and synthetic high-molecular substances such as acetyl cellulose, polyvinyl acetate, ester gum and a coumarone resin.

SUMM

. . . of alkaline-earth metals, talc, bentonite, clay and fine silica; and organic substances such as paraffin wax, polyethylene powder, pulp powder, cellulose powder and xanthone.

DETD

. . . the intermediate composite salt VII obtained in Example 7 were mixed with 1,000 ml of water, and 30 g of ammonium

aluminum sulfate (burnt alum) were added thereto. The
mixture was stirred at room temperature for 2 hours. The solid material
was separated. . .

DETD . . . the dry polyvalent metal final composite salt (final composite salt) obtained in Example 25 were kneaded with a 2-% carboxymethyl cellulose sodium salt aqueous solution. The mixture was then extruded using a disc pelletizer having a bore diameter of 1.5 mm,. .

L12 ANSWER 11 OF 12 USPATFULL on STN

AN 79:36568 USPATFULL

TI Styptic composition

IN Brown, Robert, 3249 Greenfield Dr., Marietta, GA, United States 30067 Setloff, Jerome, 6851 Roswell Rd., Apt. A 14, Atlanta, GA, United States 30328

PI US 4166108 19790828

AI US 1977-825853 19770818 (5)

RLI Continuation-in-part of Ser. No. US 1977-764405, filed on 31 Jan 1977, now abandoned

DT Utility

FS Granted

EXNAM Primary Examiner: Moyer, Donald B.

LREP Burns, Doane, Swecker & Mathis

CLMN Number of Claims: 12

ECL Exemplary Claim: 1,11,12

DRWN No Drawings

LN.CNT 374

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

- An improved styptic composition which effectively curtails bleeding while administering a soothing sensation on the injured area. The styptic composition is in the form of a stable lotion or cream having the following general formula:
 - (1) from about 2% to about 12% by weight of a long chain fatty acid;
 - (2) from about 0.4% to about 3.5% by weight of a wax filler;
 - (3) from about 1% to about 10% by weight of polyethylene glycol stearate;
 - (4) from about 1% to about 5% by weight of polyethylene glycol sorbitan beeswax; and,
 - (5) from about 0.5% to about 25% by weight of an acidic metallic salt,
 - (6) the balance being water.

Optionally, from about 1% to about 4% by weight of glycerin may be added as a humectant.

Any conventional antiseptic may also be added to the composition.

The styptic composition may be packaged in any convenient form including jars, bottles, tubes, pump applicators, aerosol canisters and can also be impregnated into pre-packaged bandages.

SUMM . . . gels which contain a variety of metallic salts in combination with a high molecular weight of polyvinyl alcohol and methyl cellulose. See, for example, U.S. Pat. No. 3,856,941. While highly acid salts are employed, the patent does not disclose the formation. . .

SUMM . . . dibasic salt as in the case of aluminum potassium sulfate. The preferred salts are those having aluminum contained therein, particularly aluminum ammonium sulfate, aluminum potassium sulfate and, as the most preferred acidic metallic

salt, aluminum sulfate. Combinations of salts may also be employed..

CLM What is claimed is:

. (5) from about 0.5% to about 25% by weight of an acidic metallic salt selected from the group consisting of aluminum chloride, aluminum sulfate, aluminum chlorohydrate, aluminum amonium sulfate, aluminum potassium sulfate, zinc chloride, zinc sulfate, zinc chlorohydrate, ferric chloride, ferric sulfate, ferric chlorohydrate and combinations thereof, said acidic metallic salt being characterized by possessing a degree of acidity sufficient to stanch. . .

. claim 1, wherein the acidic metallic salt is selected from the group consisting of aluminum sulfate, aluminum potassium sulfate and

aluminum ammonium sulfate.